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Which of the following
pairs of formulae is true
for butanoic acid?

Molecular Empirical

Formula (A)

$C_4H_8O_2$ CH_2O (B)

$C_4H_6O_2$ CH_2O (C)

$C_4H_8O_2$ C_2H_4O (D) C_4

H_8O C_2H_4O

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of 0.18 g of an organic
compound produces 0.39
6 g of carbon dioxide
and 0.216 g of water.

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QUESTION 1

SOLUTION AND
EXPLANATIONS (a)

A dative covalent or coordinate covalent bond is one in which one of the atoms supplies both the shared electrons to the covalent bond. A dative covalent bond is written $A \rightarrow B$, where the direction of

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the arrow shows the direction in which the electron pair is donated.

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group there is a change
in structure from giant

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molecular to giant

metallic and a change in

bonding from covalent

to metallic.

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